



Chapter 1: Electrons and Holes in Semiconductors



□ Course slides are prepared with the aid of the following materials:

- The lecture slides accompanying the main textbook “Hu, Chenming. *Modern semiconductor devices for integrated circuits*. Prentice Hall, 2010.”

<http://www.eecs.berkeley.edu/~hu/Book-Chapters-and-Lecture-Slides-download.html>

- The lecture slides accompanying of the Semiconductor Device Physics course offered by Dr.-Ing. Erwin Sitompul, President University, Indonesia.

<http://zitompul.wordpress.com/1-ee-lectures/2-semiconductor-device-physics/>



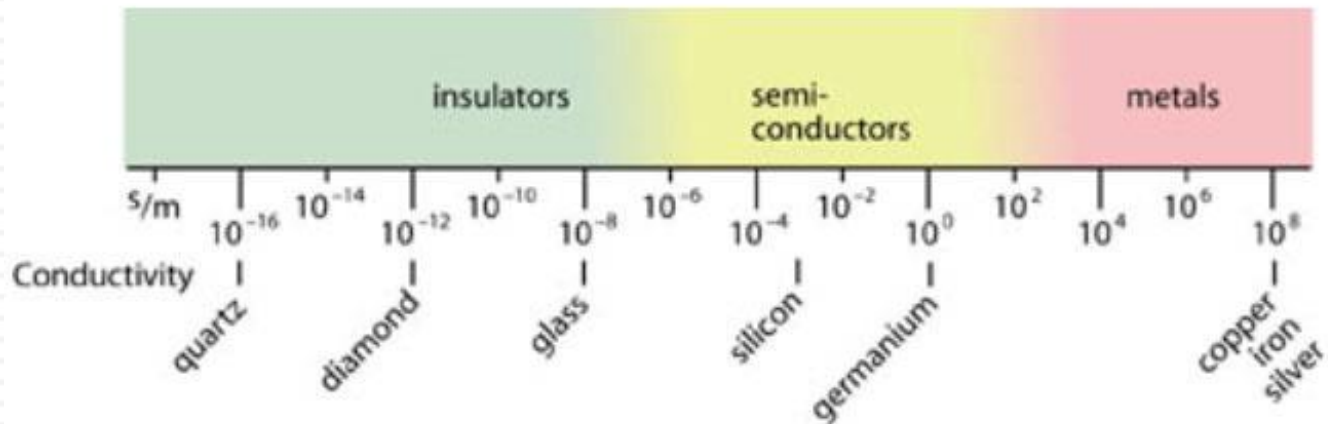
Chapter Objectives

- Provides the basic concepts and terminology for understanding semiconductors.
- Understand conduction and valence energy band, and how bandgap is formed.
- Understand carriers (electrons and holes), and doping in semiconductor
- Use the density of states and Fermi-Dirac statistics to calculate the carrier concentration



What is a Semiconductor?

- The conductivity (and at the same time the resistivity) of semiconductors lie between that of conductors and insulators.

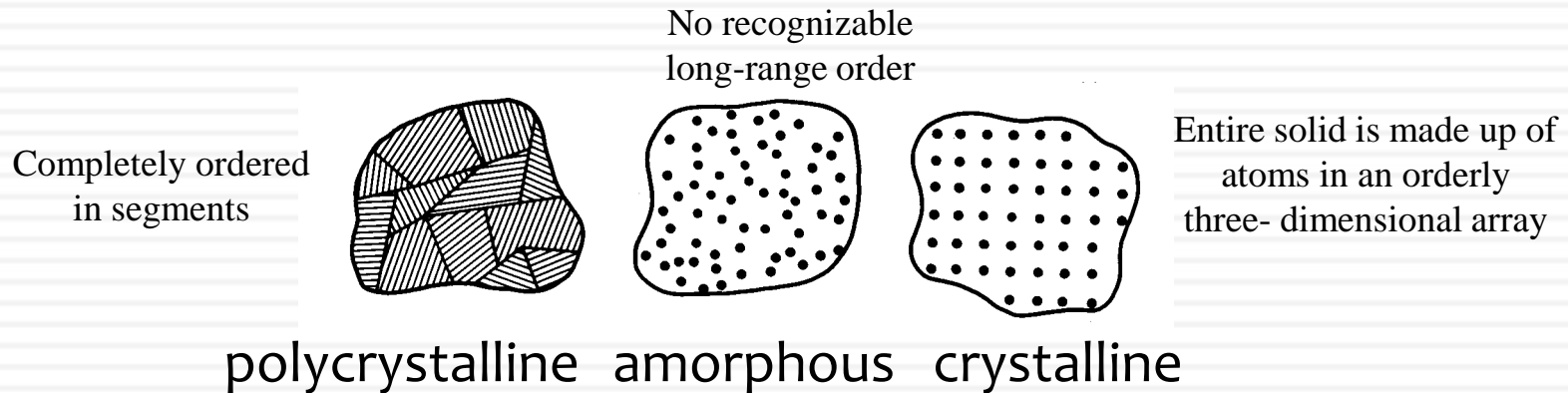


- Low resistivity → “conductor”
- High resistivity → “insulator”
- Intermediate resistivity → “semiconductor”



What is a Semiconductor?

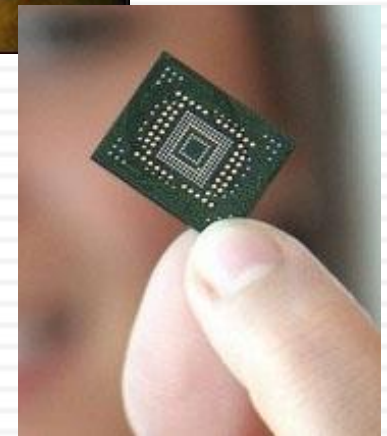
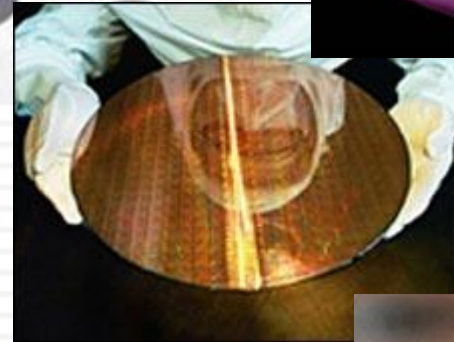
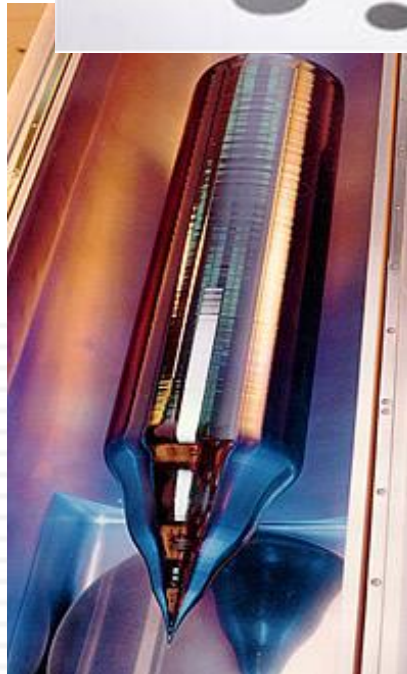
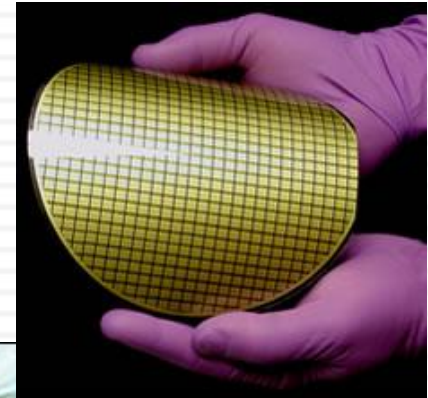
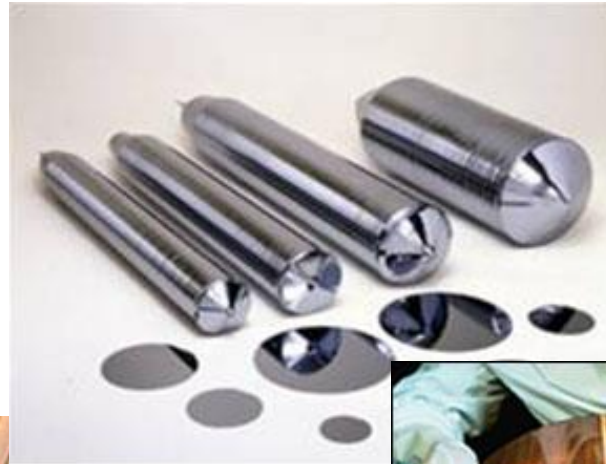
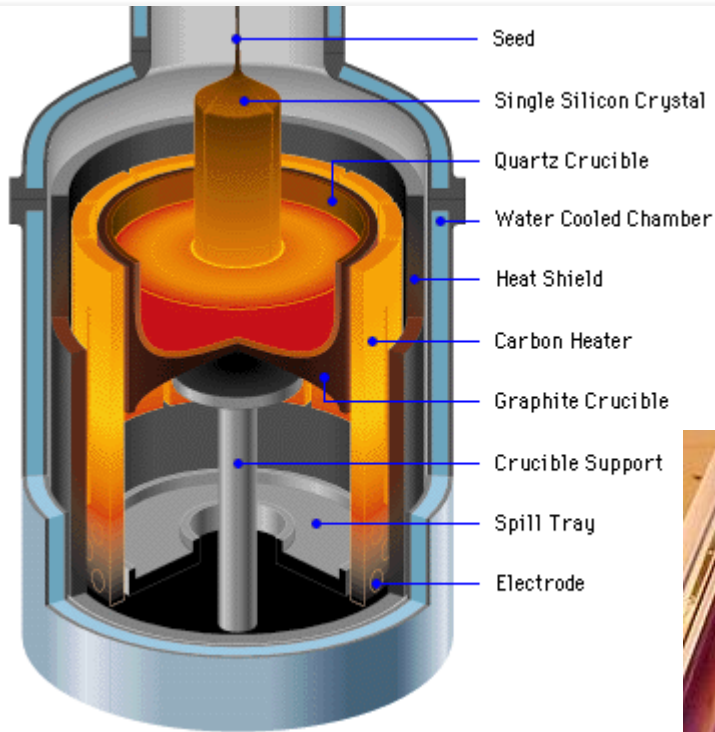
- Semiconductors are some of the purest solid materials in existence, because any trace of impurity atoms called “dopants” can change the electrical properties of semiconductors drastically.



- Most devices fabricated today employ crystalline semiconductors.



Crystal Growth Until Device Fabrication





Semiconductor Materials

Elemental: Si, Ge, C

Compound: IV-IV SiC
III-V GaAs, GaN
II-VI CdSe

Alloy: $\text{Si}_{1-x}\text{Ge}_x$
 $\text{Al}_x\text{Ga}_{1-x}\text{As}$

As : Arsenic

Cd : Cadmium

Se : Selenium

Ga : Gallium

11 12 13 14 15 16 17 18

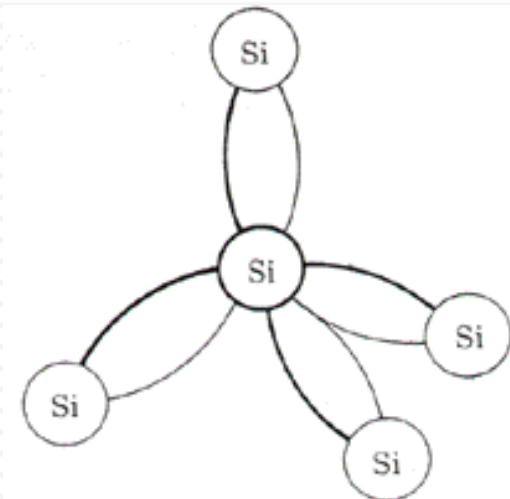
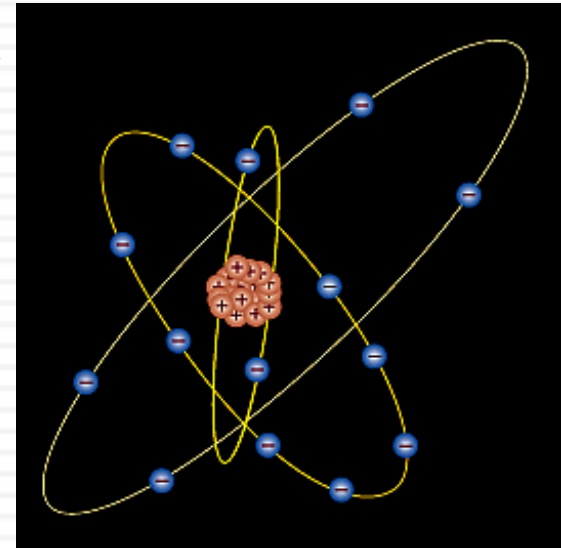
							2 He
		5 B	6 C	7 N	8 O	9 F	10 Ne
		13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
111 Rg	112 Uub	113 Uut	114 Uuq	115 Uup	116 Uuh	117 Uus	118 Uuo

65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No



The Silicon Atom

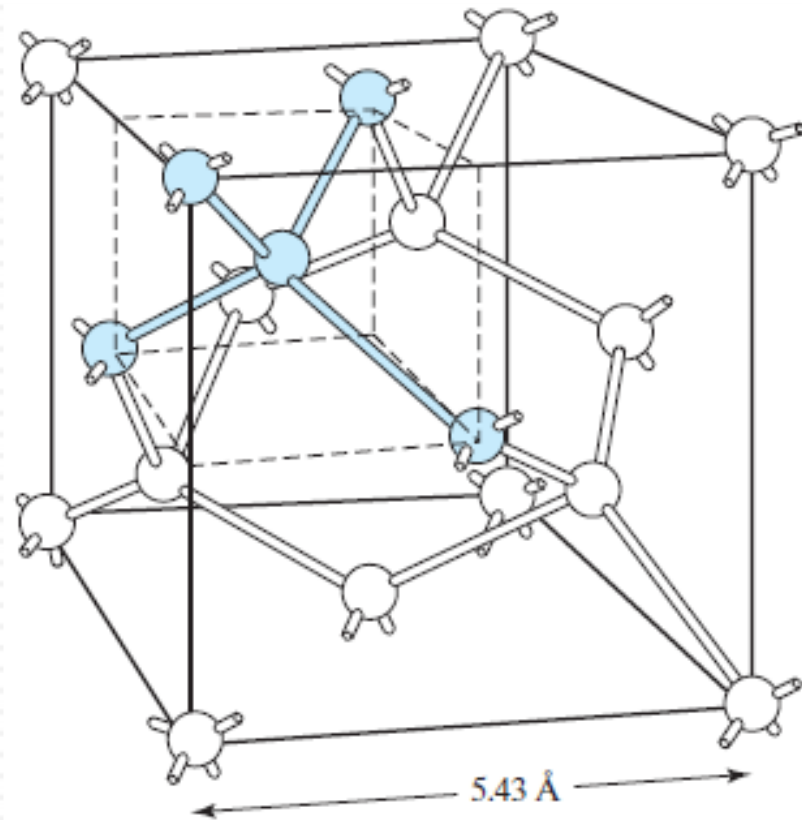
- 14 electrons occupying the first 3 energy levels:
 - 1s, 2s, 2p orbitals are filled by 10 electrons.
 - 3s, 3p orbitals filled by 4 electrons.
- To minimize the overall energy, the 3s and 3p orbitals hybridize to form four tetrahedral 3sp orbital.
- Each has one electron and is capable of forming a bond with a neighboring atom.





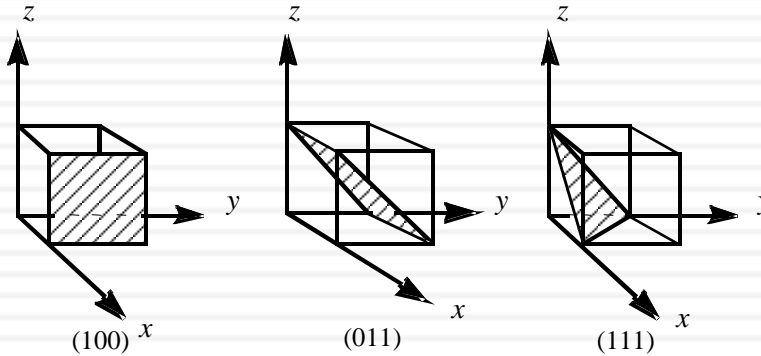
1.1 Silicon Crystal Structure

- **Unit cell** of silicon crystal is cubic.
- **Each Si atom has 4 nearest neighbors.**
- **Each cell contains:**
 - 8 corner atoms
 - 6 face atoms
 - 4 interior atoms
- **Exercise**
How Many Silicon Atoms per cm^{-3} ?

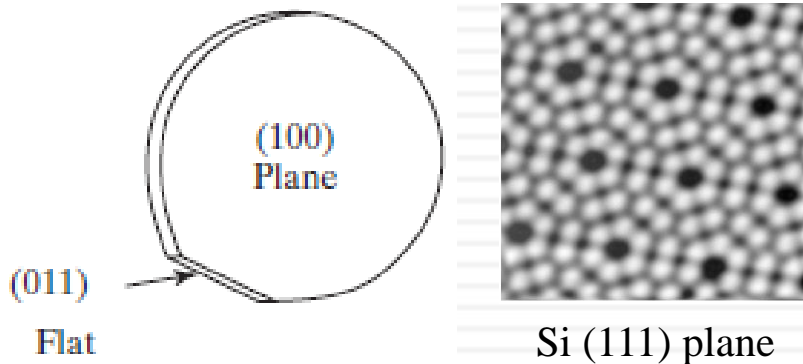




Silicon Wafers and Crystal Planes (Miller Indices)



- The standard notation for crystal planes is based on the cubic unit cell.

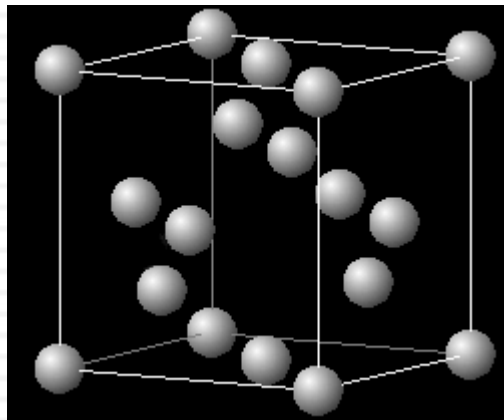


- Silicon wafers are usually cut along the (100) plane with a flat or notch to help orient the wafer during IC fabrication.

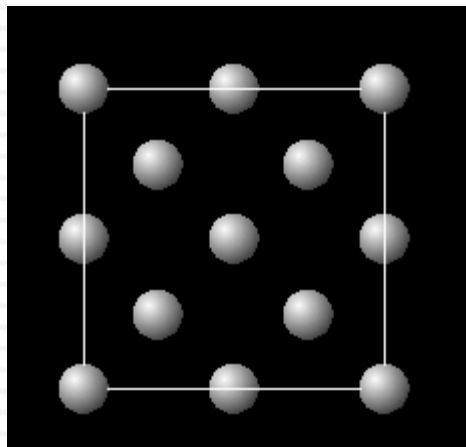


Crystallographic Planes of Si

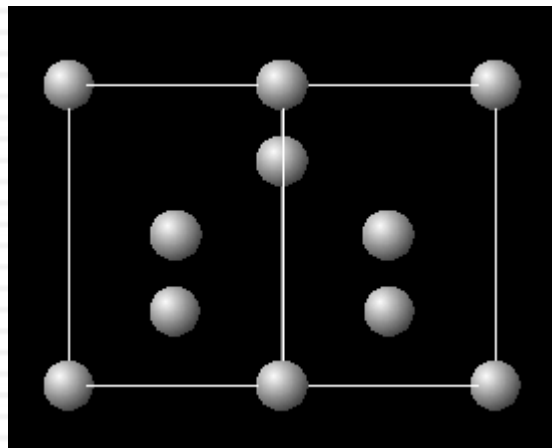
Unit cell:



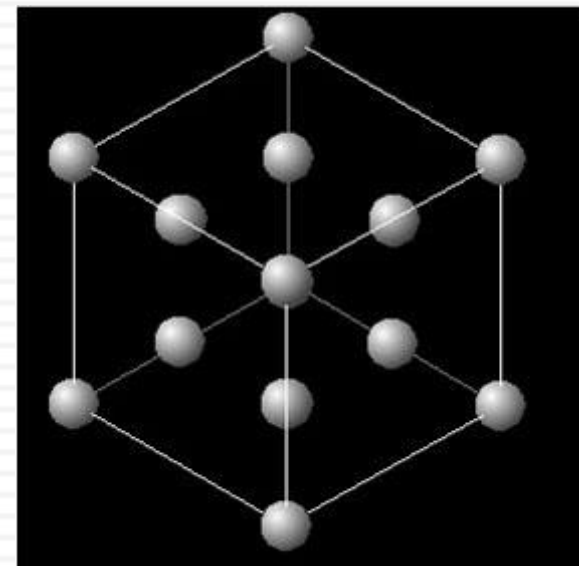
View in $\langle 100 \rangle$ direction



View in $\langle 110 \rangle$ direction

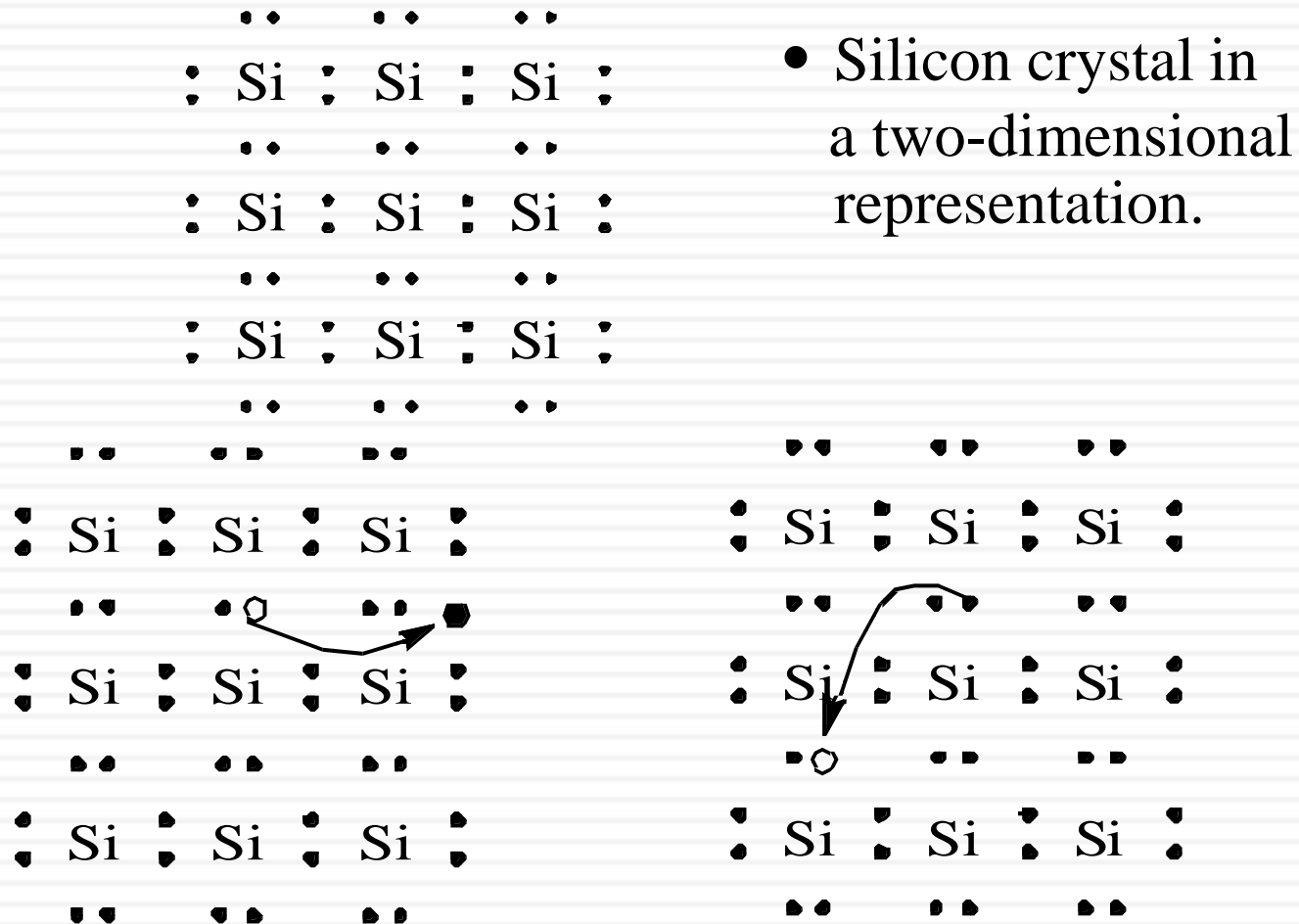


View in $\langle 111 \rangle$ direction





1.2 Bond Model of Electrons and Holes



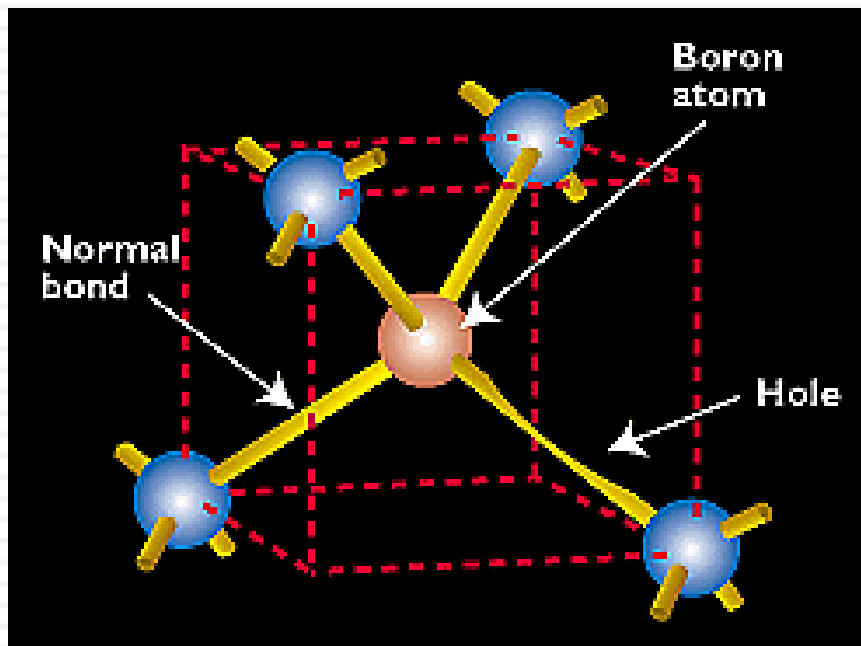
- When an electron breaks loose and becomes a *conduction electron*, a *hole* is also created.



Doping - Manipulation of Carrier Numbers

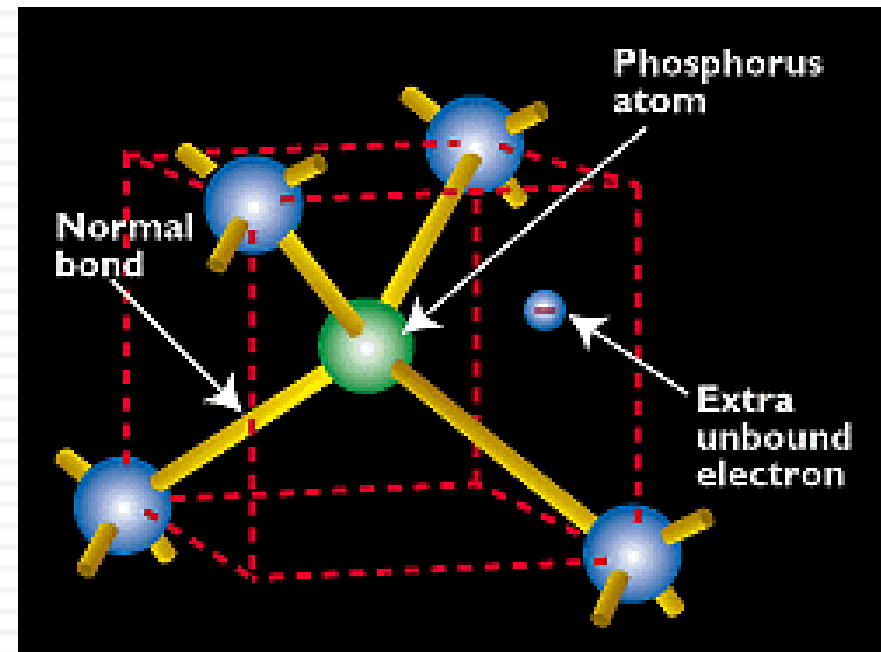
- By substituting an Si atom with a special impurity atom (elements from **Group III** or **Group V**), a hole or conduction electron can be created.

Acceptors: B, Ga, In, Al



Boron, Gallium Indium, Aluminum

Donors: P, As, Sb

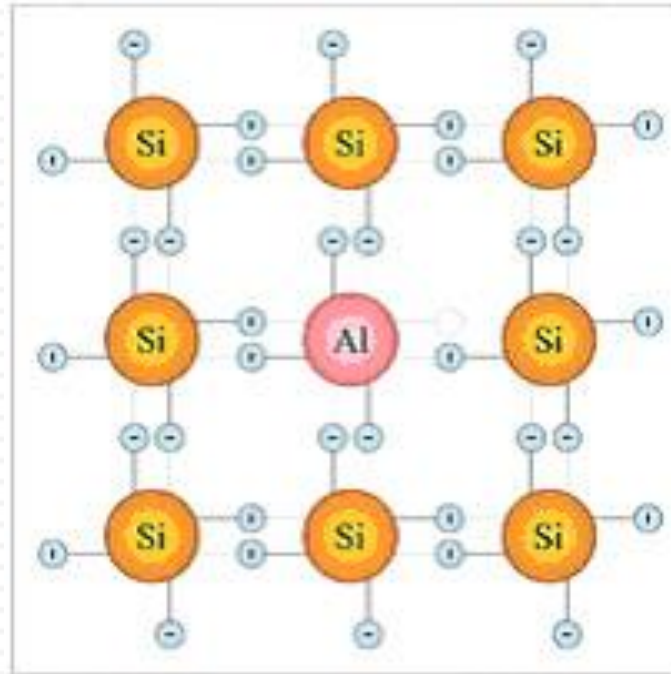


Phosphorus, Arsenic, Antimony



Doping Silicon with Acceptors

- **Example:** Aluminum atom is doped into the Si crystal.

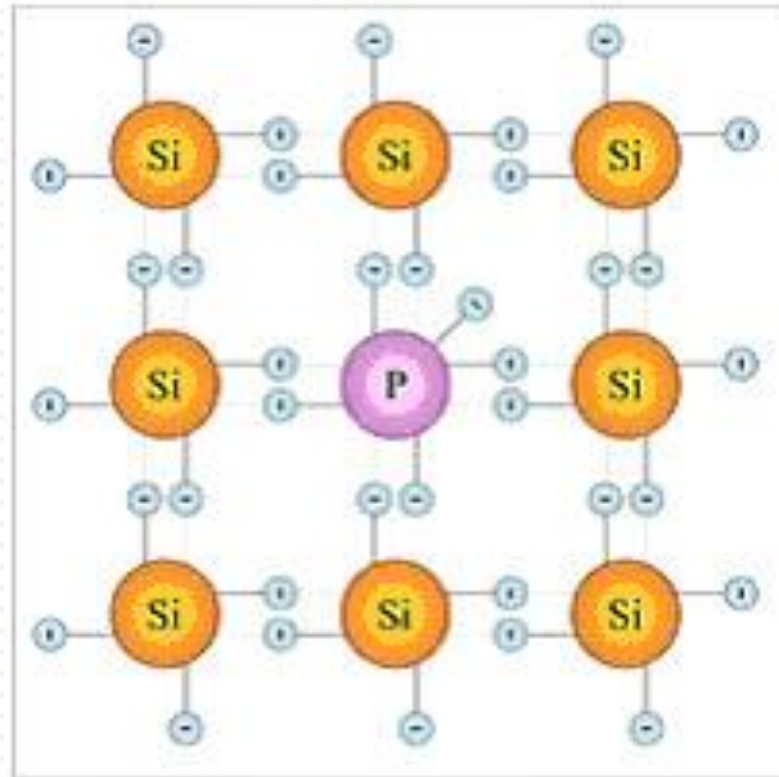


- The **Al** atom accepts an electron from a neighboring **Si** atom, resulting in a missing bonding electron, or “hole”.
- The hole is free to roam around the **Si** lattice, and as a moving positive charge, the hole carries current.



Doping Silicon with Donors

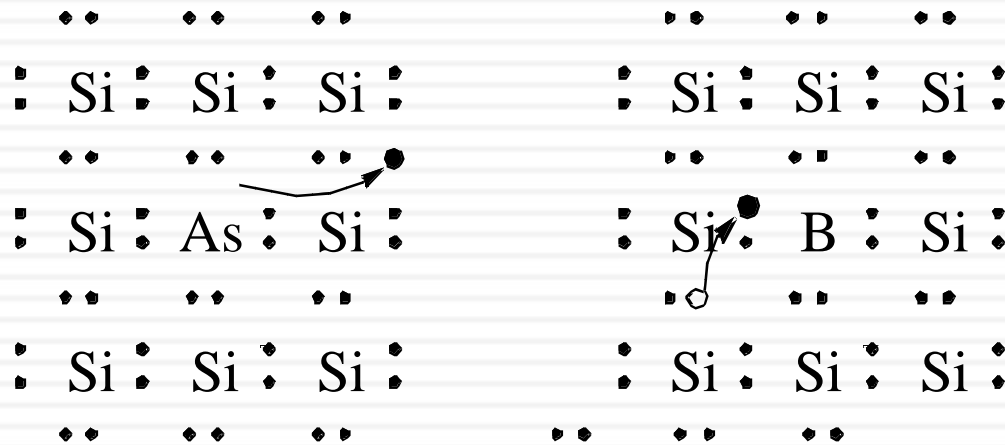
- **Example:** Phosphorus atom is doped into the Si crystal.



- The loosely bounded fifth valence electron of the **P** atom can “break free” easily and becomes a mobile conducting electron.
- This electron contributes in current conduction.



Dopants in Silicon



- As, a Group V element, introduces conduction electrons and creates ***N-type silicon***, and is called a ***donor***.
- B, a Group III element, introduces holes and creates ***P-type silicon***, and is called an ***acceptor***.
- Donors and acceptors are known as dopants. Dopant ionization energy $\sim 50\text{meV}$ (very low).

$$\text{Hydrogen: } E_{ion} = \frac{m_0 q^4}{8e_0^2 h^2} = 13.6 \text{ eV}$$

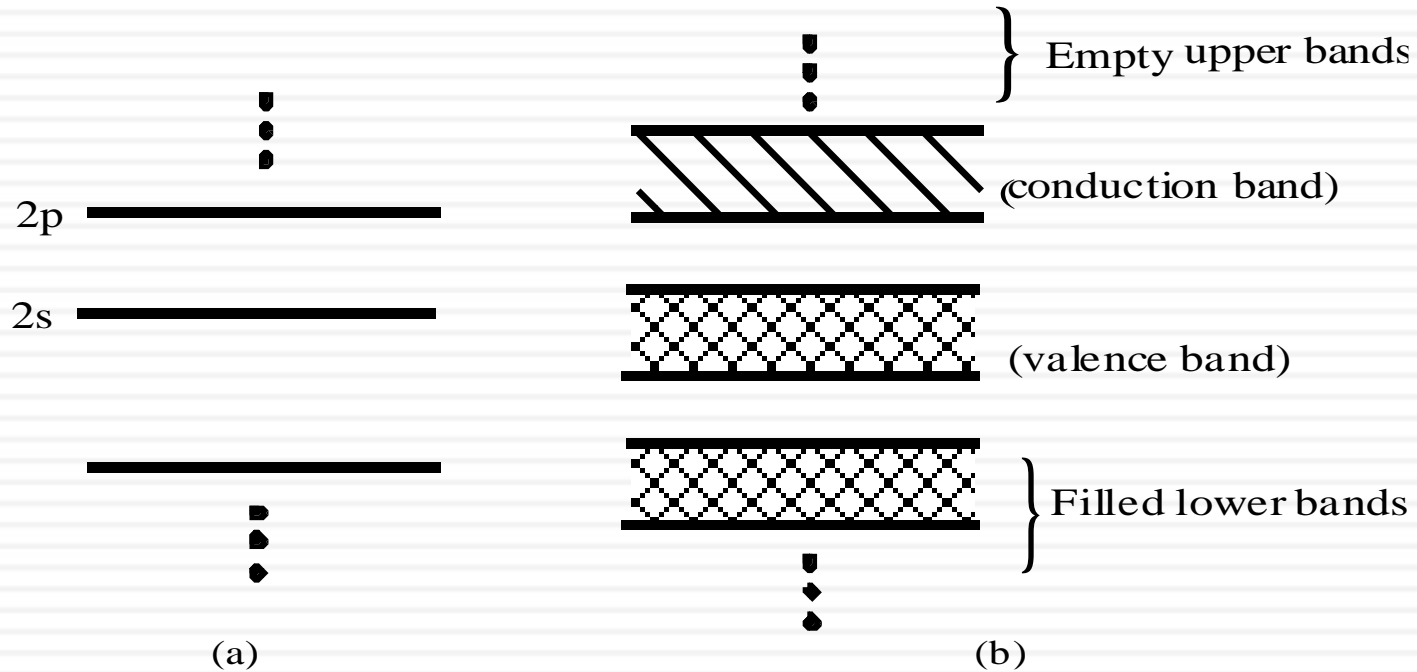


Carrier-Related Terminology

- **Donor:** impurity atom that increases n (conducting electron).
Acceptor: impurity atom that increases p (hole).
- **n -type material:** contains more electrons than holes.
 p -type material: contains more holes than electrons.
- **Majority carrier:** the most abundant carrier.
Minority carrier: the least abundant carrier.
- **Intrinsic semiconductor:** undoped semiconductor $n = p = n_i$.
Extrinsic semiconductor: doped semiconductor.



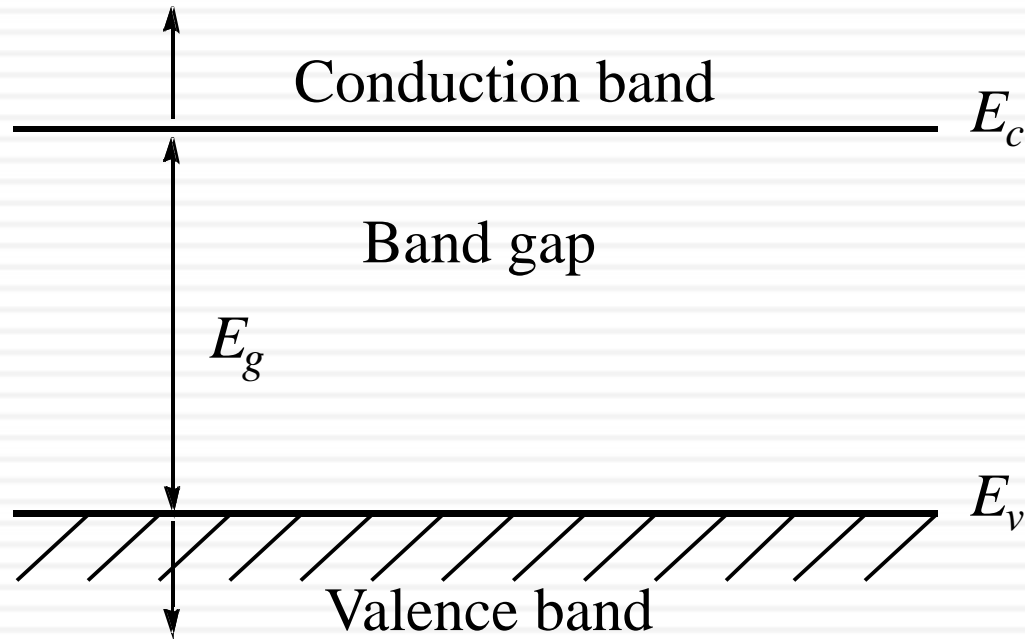
1.3 Energy Band Model



- Energy states of Si atom (a) expand into energy bands of Si crystal (b).
- The lower bands are filled and higher bands are empty in a semiconductor.
- The highest filled band is the **valence band**.
- The lowest empty band is the **conduction band**



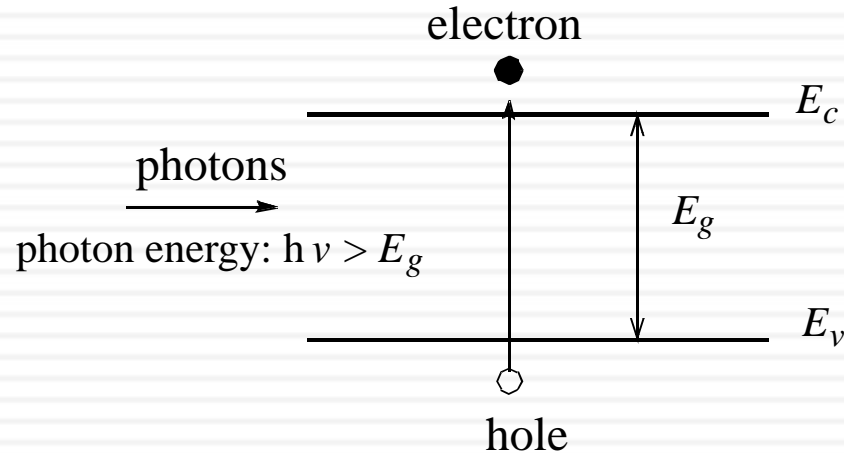
1.3.1 Energy Band Diagram



- **Energy band diagram** shows the bottom edge of conduction band, E_c , and top edge of valence band, E_v .
- E_c and E_v are separated by the **band gap energy, E_g** .



Measuring the Band Gap Energy by Light Absorption



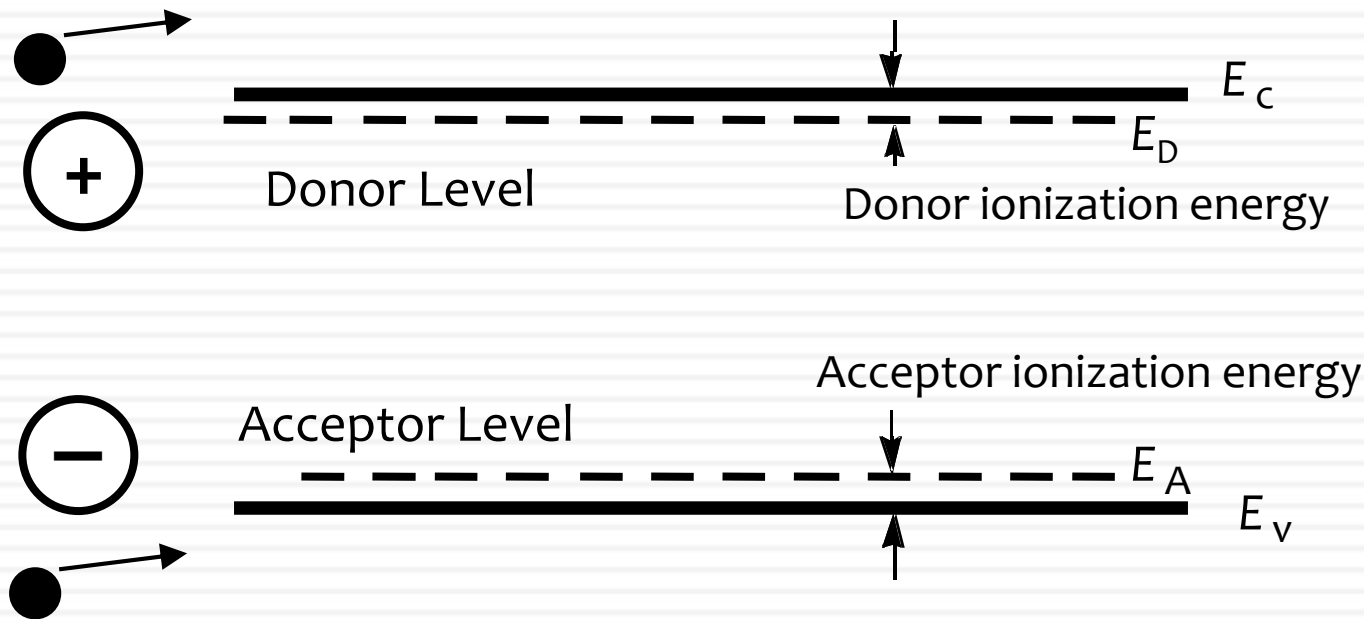
- E_g can be determined from the minimum energy (hn) of photons that are absorbed by the semiconductor.

Bandgap energies of selected semiconductors

Semi-conductor	InSb	Ge	Si	GaAs	GaP	ZnSe	Diamond
E_g (eV)	0.18	0.67	1.12	1.42	2.25	2.7	6



Donor / Acceptor Levels (Band Model)



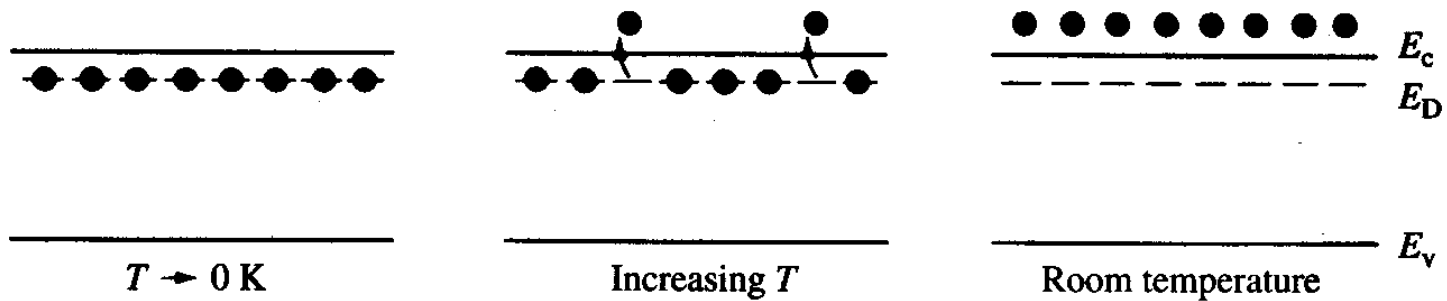
**Ionization energy of selected donors and acceptors
in Silicon ($E_G = 1.12$ eV)**

	Donors			Acceptors		
	Sb	P	As	B	Al	In
Ionization energy of dopant $E_C - E_D$ or $E_A - E_V$ (meV)	39	45	54	45	67	160

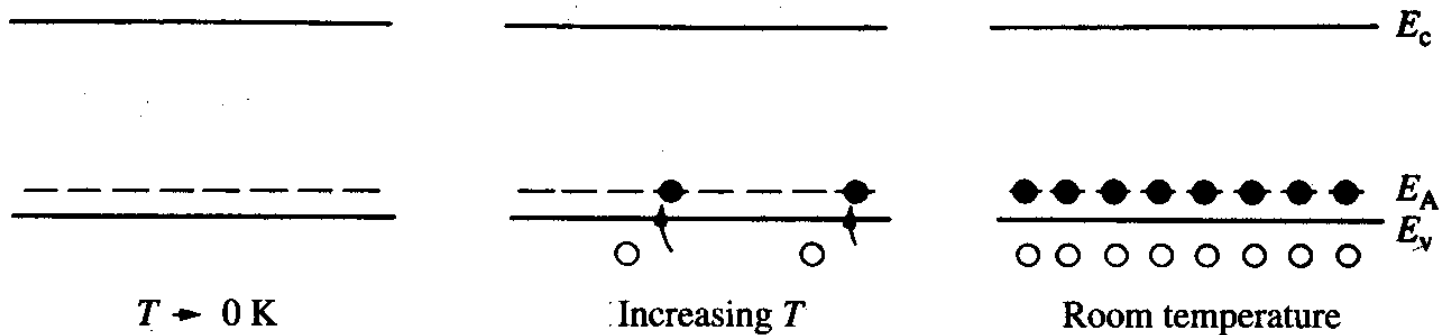


Dopant Ionization (Band Model)

Donor atoms

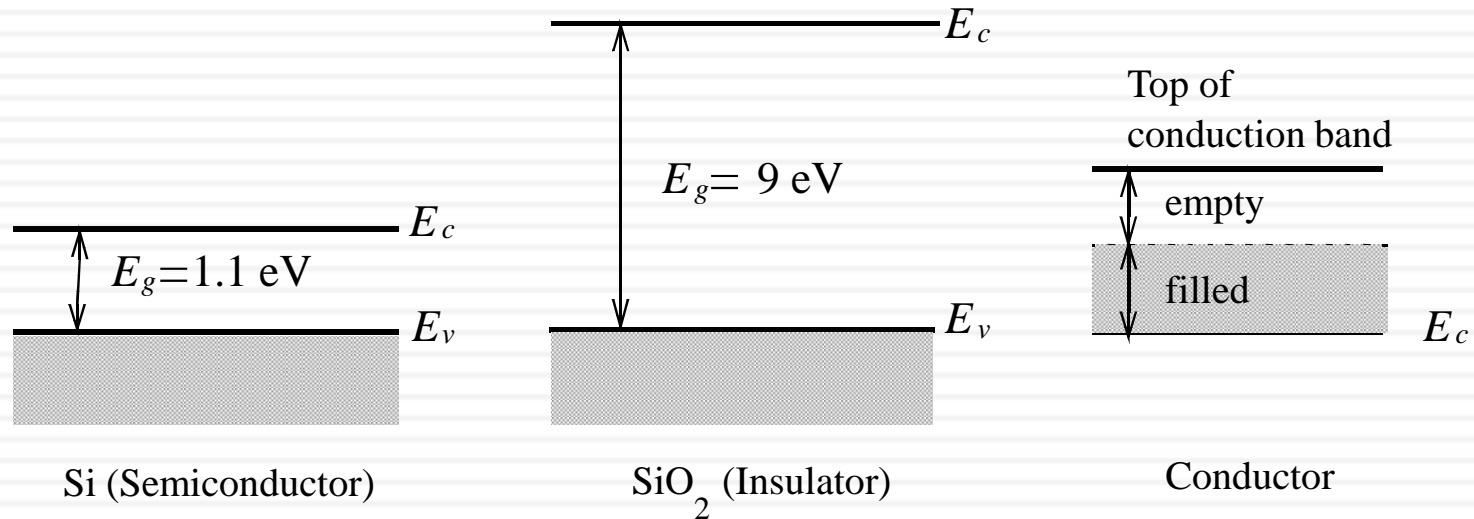


Acceptor atoms





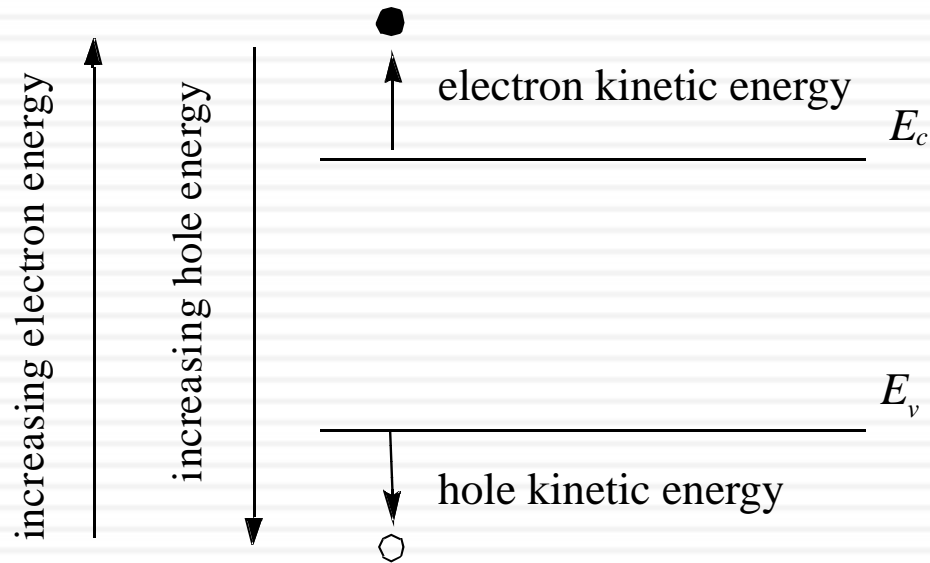
1.4 Semiconductors, Insulators, and Conductors



- Totally filled bands and totally empty bands do not allow current flow. (Just as there is no motion of liquid in a totally filled or totally empty bottle.)
- Metal conduction band is half-filled.
- Semiconductors have lower E_g 's than insulators and can be doped.



1.5 Electrons and Holes



- Both electrons and holes tend to seek their lowest energy positions.
- Electrons tend to fall in the energy band diagram.
- Holes float up like bubbles in water



1.5.1 Effective Mass

The electron wave function is the solution of the three dimensional Schrodinger wave equation

$$-\frac{\hbar^2}{2m_0} \nabla^2 \psi + V(r) \psi = E \psi$$

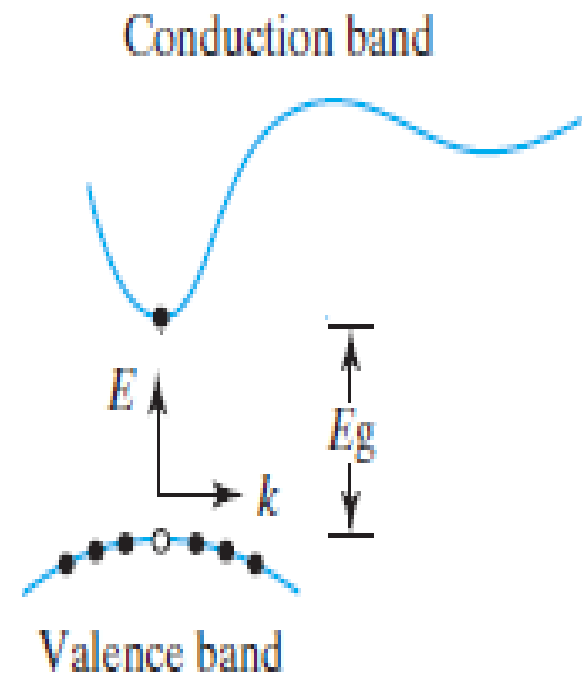
The solution is of the form $\exp(\pm \mathbf{k} \cdot \mathbf{r})$

k = wave vector = $2\pi/\text{electron wavelength}$

For each k , there is a corresponding E .

$$\text{acceleration} = -\frac{q\varepsilon}{\hbar^2} \frac{d^2 E}{dk^2} = \frac{F}{m}$$

$$\text{effective mass} \equiv \frac{\hbar^2}{d^2 E / dk^2}$$





1.5.1 Effective Mass

In an electric field, \mathcal{E} , an electron or a hole accelerates.

$$a = \frac{-q\mathcal{E}}{m_n} \quad \text{electrons}$$

$$a = \frac{q\mathcal{E}}{m_p} \quad \text{holes}$$

Electron and hole effective masses

	Si	Ge	GaAs	InAs	AlAs
m_n/m_0	0.26	0.12	0.068	0.023	2
m_p/m_0	0.39	0.3	0.5	0.3	0.3



1.5.2 How to Measure the Effective Mass

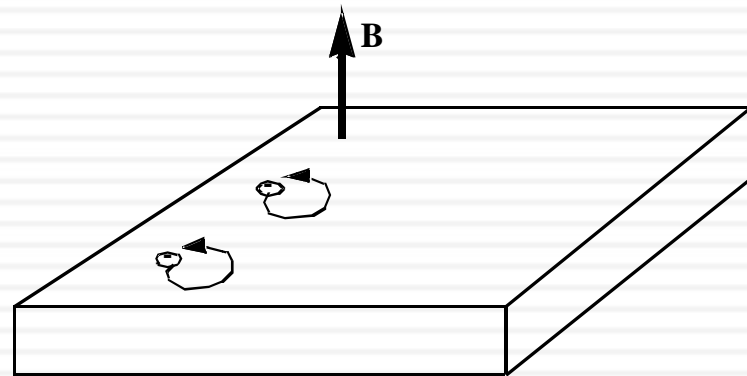
Cyclotron Resonance Technique

Centripetal force = Lorentzian force

$$\frac{m_n v^2}{r} = qvB$$

$$v = \frac{qBr}{m_n}$$

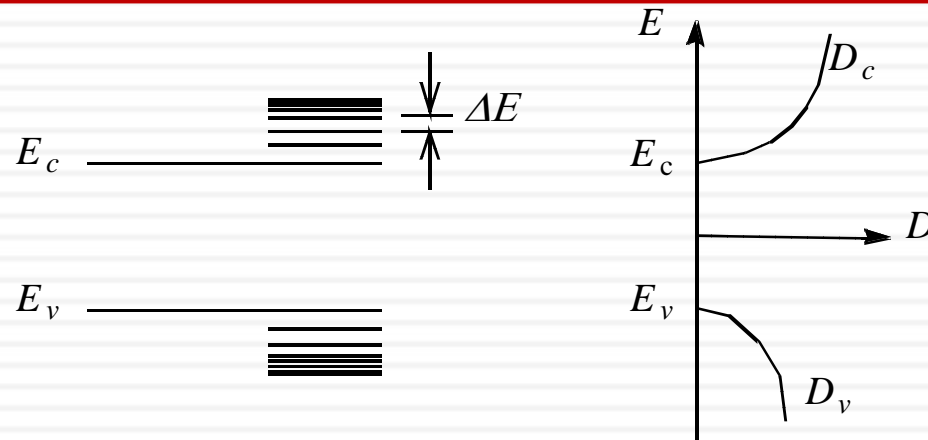
$$f_{cr} = \frac{v}{2\pi r} = \frac{qB}{2\pi m_n}$$



- f_{cr} is the Cyclotron resonance frequency.
- It is independent of v and r .
- Electrons strongly absorb microwaves of that frequency.
- By measuring f_{cr} , m_n can be found.



1.6 Density of States



$$D_c(E) \equiv \frac{\text{number of states in } \Delta E}{\Delta E \cdot \text{volume}} \left(\frac{1}{\text{eV} \cdot \text{cm}^3} \right)$$

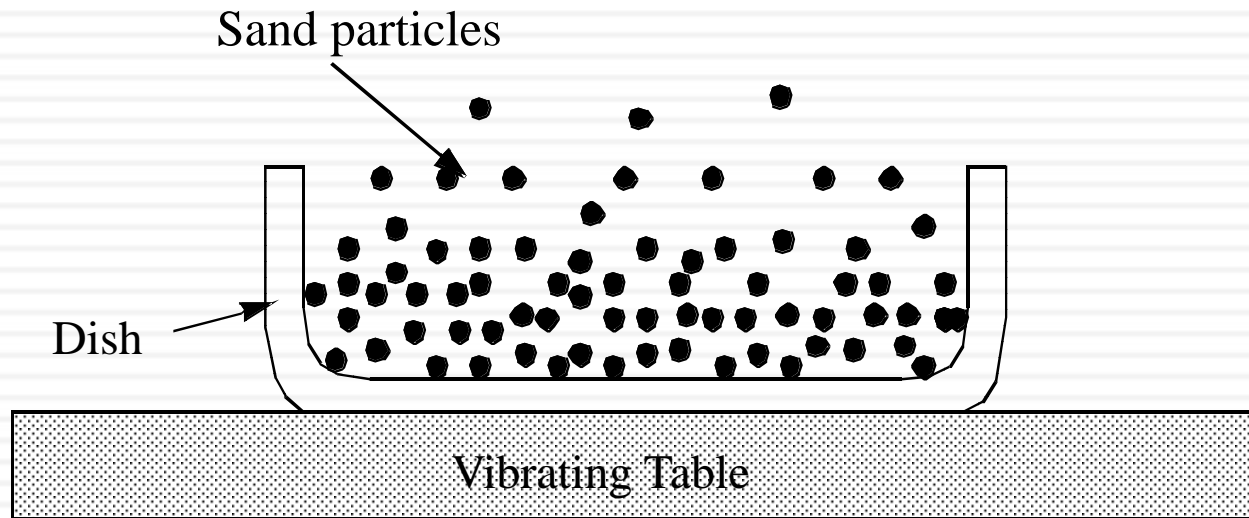
$$D_c(E) \equiv \frac{8\pi m_n \sqrt{2m_n(E - E_c)}}{h^3}$$

$$D_v(E) \equiv \frac{8\pi m_p \sqrt{2m_p(E_v - E)}}{h^3}$$



1.7 Thermal Equilibrium and the Fermi Function

1.7.1 An Analogy for Thermal Equilibrium



- There is a certain probability for the electrons in the conduction band to occupy high-energy states under the agitation of thermal energy.



1.7.2 Fermi Function–The Probability of an Energy State Being Occupied by an Electron

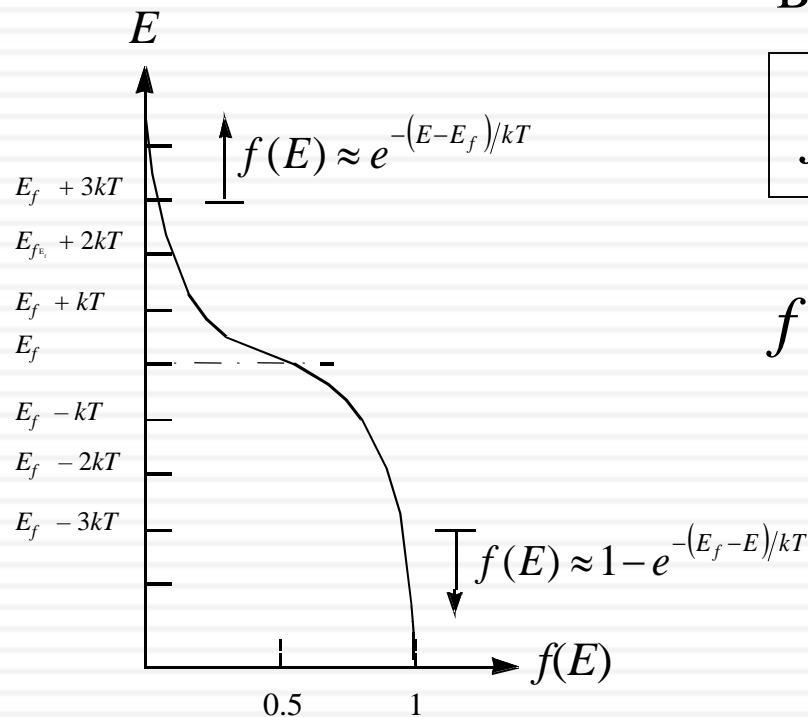
$$f(E) = \frac{1}{1 + e^{(E-E_f)/kT}}$$

E_f is called the *Fermi energy* or the *Fermi level*.

Boltzmann approximation:

$$f(E) \approx e^{-(E-E_f)/kT} \quad E - E_f \gg kT$$

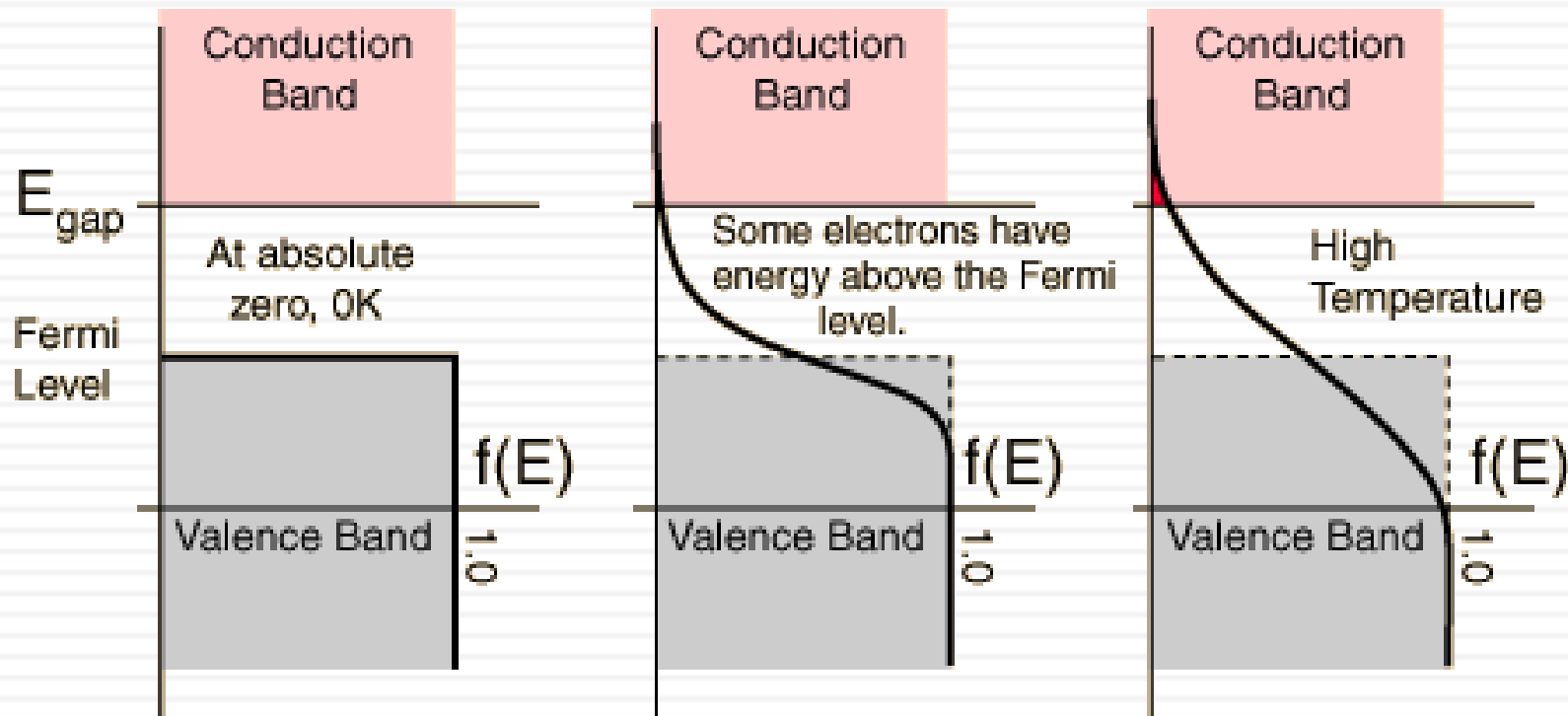
$$f(E) \approx 1 - e^{-(E_f-E)/kT} \quad E - E_f \ll -kT$$



Remember: there is only one Fermi-level in a system at equilibrium.



Effect of Temperature on $f(E)$

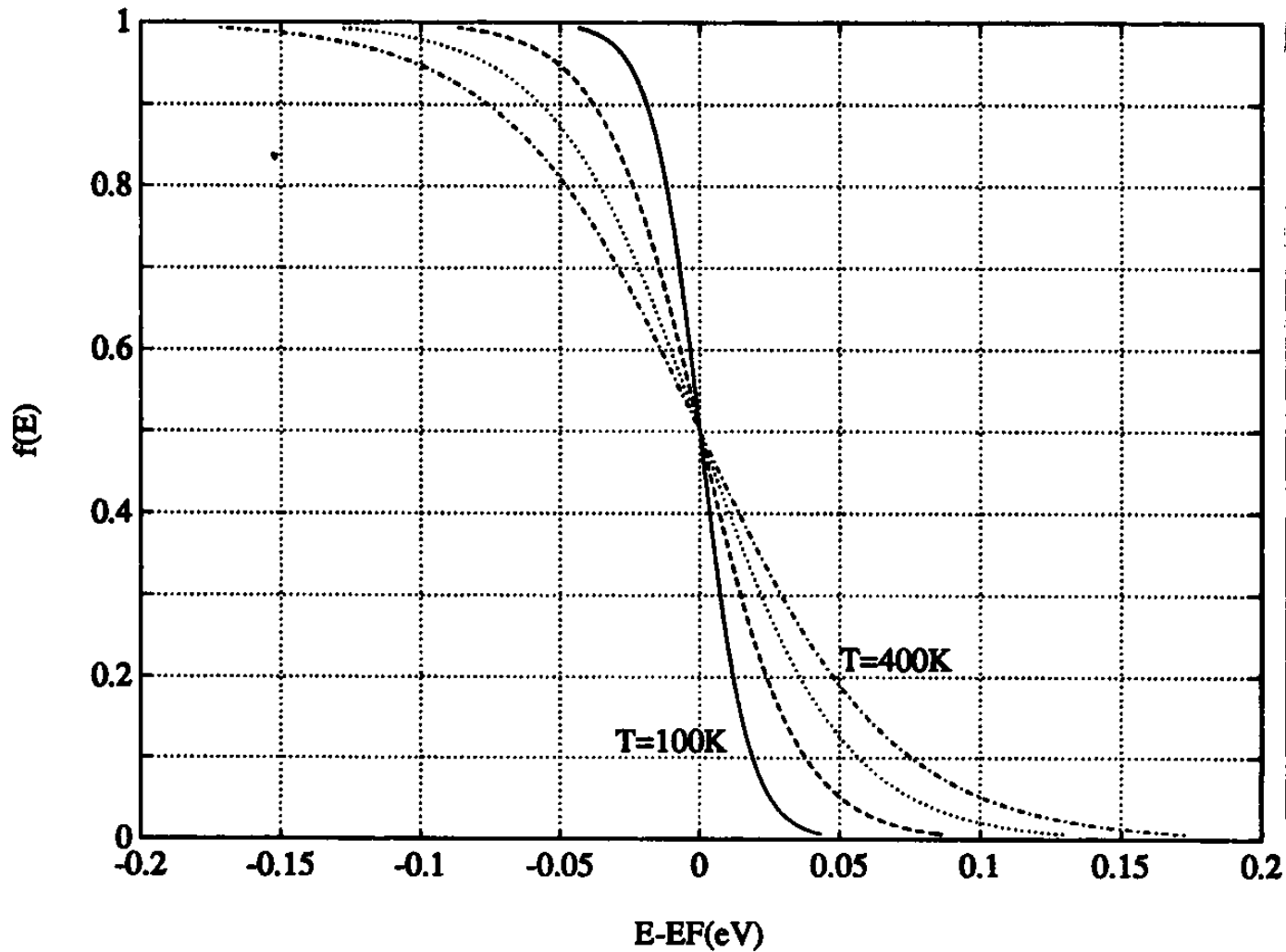


No electrons can be above the valence band at 0K, since none have energy above the Fermi level and there are no available energy states in the band gap.

At high temperatures, some electrons can reach the conduction band and contribute to electric current.



Effect of Temperature on $f(E)$

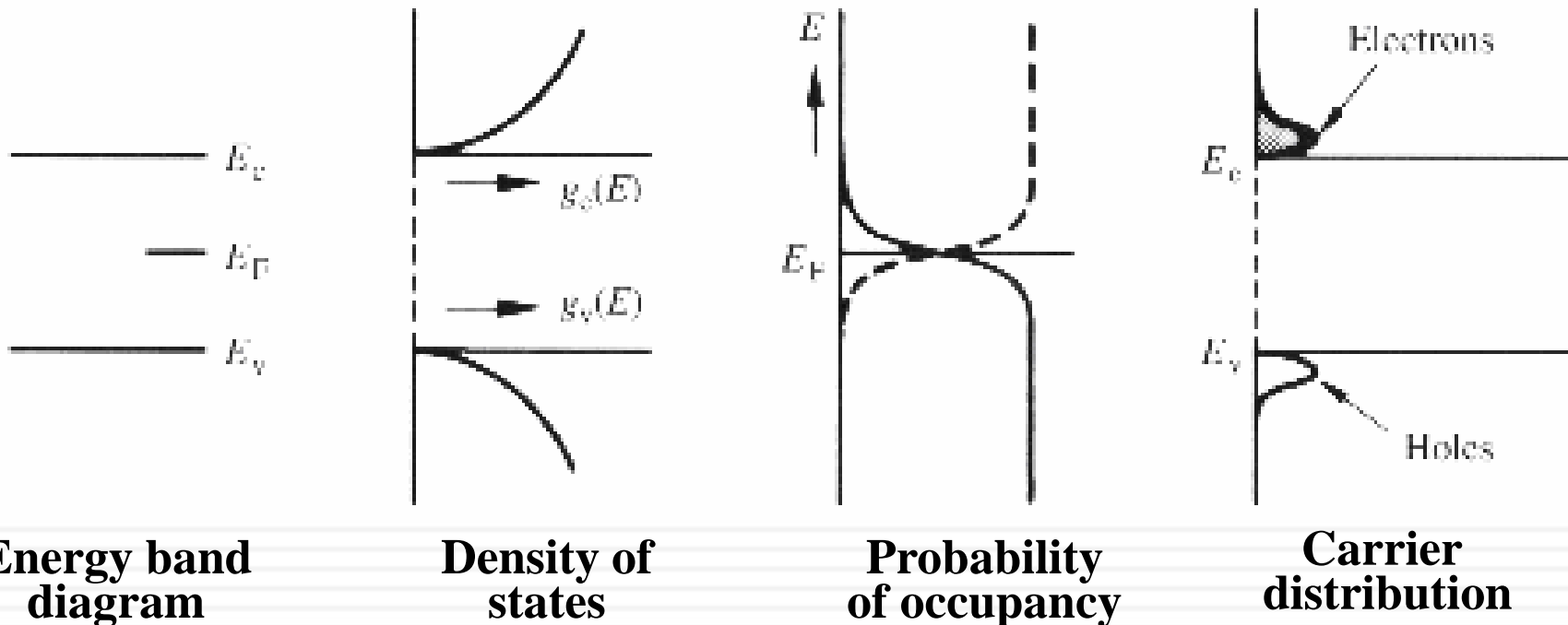




Equilibrium Distribution of Carriers - Intrinsic

- $n(E)$ is obtained by multiplying $g_c(E)$ and $f(E)$,
 $p(E)$ is obtained by multiplying $g_v(E)$ and $1-f(E)$.

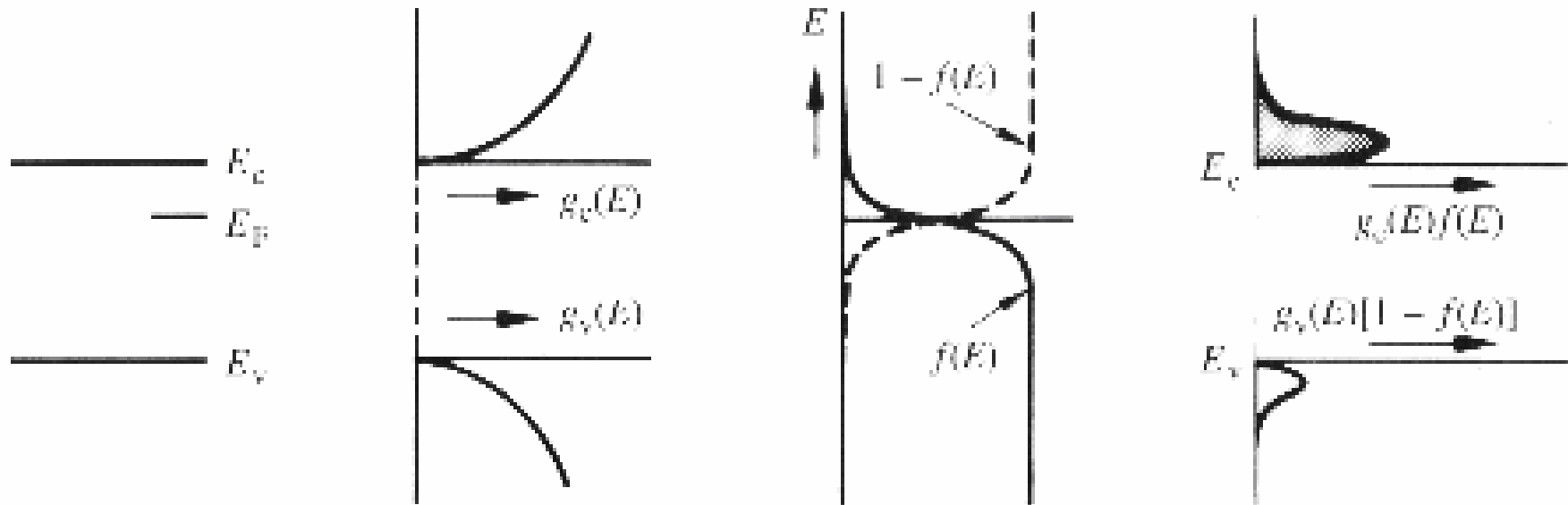
- **Intrinsic semiconductor material**





Equilibrium Distribution of Carriers – n-type

■ n-type semiconductor material



Energy band diagram

Density of States

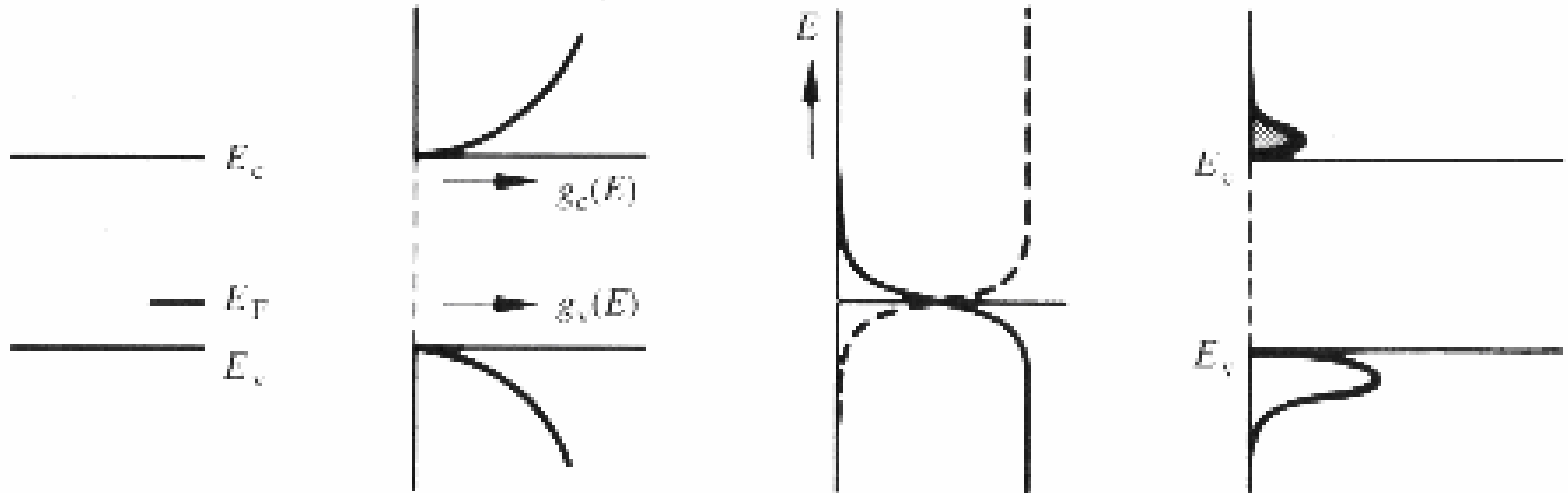
Probability of occupancy

Carrier distribution



Equilibrium Distribution of Carriers – p-type

■ p-type semiconductor material



**Energy band
diagram**

**Density of
States**

**Probability
of occupancy**

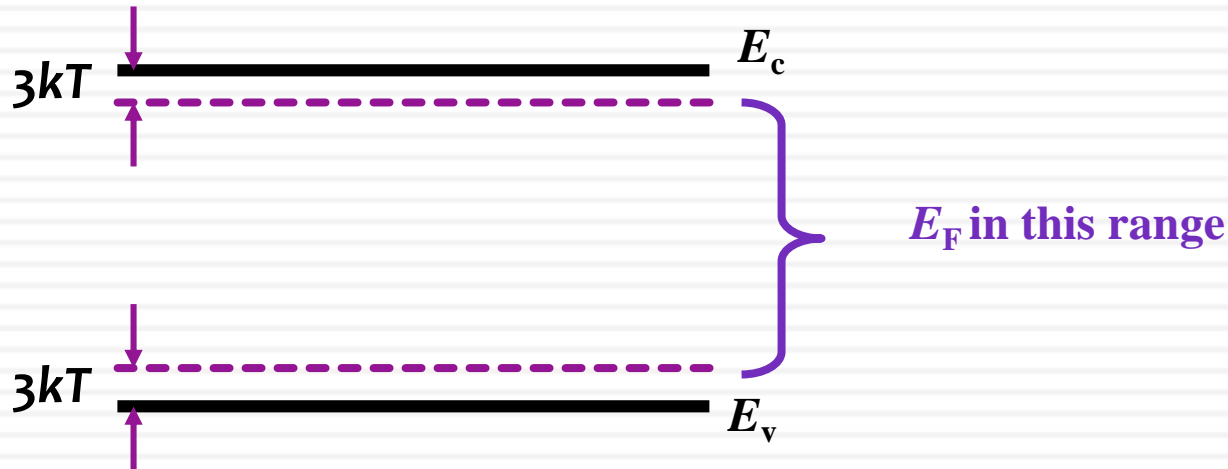
**Carrier
distribution**



Nondegenerately Doped Semiconductor

- The expressions for n and p will now be derived in the range where the Boltzmann approximation can be applied:

$$E_v + 3kT \leq E_F \leq E_c - 3kT$$



- The semiconductor is said to be **nondegenerately doped (lightly doped)** in this case.



Degenerately Doped Semiconductor

- If a semiconductor is very heavily doped, the Boltzmann approximation is not valid.
- For Si at $T = 300$ K,
 - $E_C - E_F < 3kT$ if $N_D > 1.6 \times 10^{18} \text{ cm}^{-3}$
 - $E_F - E_V < 3kT$ if $N_A > 9.1 \times 10^{17} \text{ cm}^{-3}$
- The semiconductor is said to be **degenerately doped (heavily doped)** in this case.
 - $N_D = \text{total number of donor atoms/cm}^3$
 - $N_A = \text{total number of acceptor atoms/cm}^3$



Important Constants

- Electronic charge, $q = 1.6 \times 10^{-19} \text{ C}$
- Permittivity of free space, $\epsilon_0 = 8.854 \times 10^{-12} \text{ F/m}$
- Boltzmann constant, $k = 8.62 \times 10^{-5} \text{ eV/K}$
- Planck constant, $h = 4.14 \times 10^{-15} \text{ eV}\cdot\text{s}$, $\hbar = h/2\pi$
- Free electron mass, $m_0 = 9.1 \times 10^{-31} \text{ kg}$
- Thermal energy, $kT = 0.02586 \text{ eV}$ (at 300 K)
- Thermal voltage, $kT/q = 0.02586 \text{ V}$ (at 300 K)
- Silicon energy band gap, $E_G = 1.12 \text{ eV}$
- Intrinsic Si carrier concentration $n_i = 1 \times 10^{10} \text{ cm}^{-3}$ (at 300 K)



1.8 Electron and Hole Concentrations

1.8.1 Derivation of n and p from $D(E)$ and $f(E)$

$$n = \int_{E_c}^{\text{top of conduction band}} f(E) D_c(E) dE$$

$$n = \frac{8\pi m_n \sqrt{2m_n}}{h^3} \int_{E_c}^{\infty} \sqrt{E - E_c} e^{-(E - E_f)/kT} dE$$

$$= \frac{8\pi m_n \sqrt{2m_n}}{h^3} e^{-(E_c - E_f)/kT} \int_0^{E - E_c} \sqrt{E - E_c} e^{-(E - E_c)/kT} d(E - E_c)$$

- Similarly p can be derived

$$p = \int_{E_{\text{bottom}}}^{E_v} D_v(E) [1 - f(E)] dE$$



Electron and Hole Concentrations

$$n = N_c e^{-(E_c - E_f)/kT}$$

$$N_c \equiv 2 \left[\frac{2\pi m_n kT}{h^2} \right]^{3/2}$$

$$p = N_v e^{-(E_f - E_v)/kT}$$

$$N_v \equiv 2 \left[\frac{2\pi m_p kT}{h^2} \right]^{3/2}$$

N_c is called the *effective density of states (of the conduction band)*.

N_v is called the *effective density of states of the valence band*.

Remember: the closer E_f moves up to N_c , the larger n is; the closer E_f moves down to E_v , the larger p is.

For Si, $N_c = 2.8 \times 10^{19} \text{ cm}^{-3}$ and $N_v = 1.04 \times 10^{19} \text{ cm}^{-3}$.



1.8.2 The Fermi Level and Carrier Concentrations

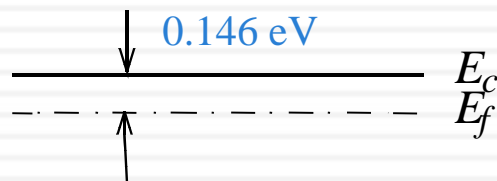
Example: Where is E_f for $n = 10^{17} \text{ cm}^{-3}$? And for $p = 10^{14} \text{ cm}^{-3}$?

Solution: (a) $n = N_c e^{-(E_c - E_f)/kT}$

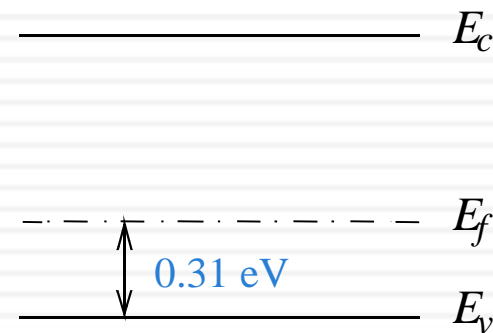
$$E_c - E_f = kT \ln(N_c/n) = 0.026 \ln(2.8 \times 10^{19} / 10^{17}) = 0.146 \text{ eV}$$

(b) For $p = 10^{14} \text{ cm}^{-3}$, from Eq.(1.8.8),

$$E_f - E_v = kT \ln(N_v/p) = 0.026 \ln(1.04 \times 10^{19} / 10^{14}) = 0.31 \text{ eV}$$



(a)



(b)



1.8.3 The np Product and the Intrinsic Carrier Concentration

Multiply $n = N_c e^{-(E_c - E_f)/kT}$ and $p = N_v e^{-(E_f - E_v)/kT}$

$$np = N_c N_v e^{-(E_c - E_v)/kT} = N_c N_v e^{-E_g/kT}$$

$$np = n_i^2$$

$$n_i = \sqrt{N_c N_v} e^{-E_g/2kT}$$

- In an intrinsic (undoped) semiconductor, $n = p = n_i$.
- n_i is the *intrinsic carrier concentration*, $\sim 10^{10} \text{ cm}^{-3}$ for Si.



Alternative Expressions: $n(n_i, E_i)$ and $p(n_i, E_i)$

- In an intrinsic semiconductor, $n = p = n_i$ and $E_F = E_i$, where E_i denotes the intrinsic Fermi level.

$$n = N_C e^{(E_F - E_c)/kT}$$

$$n_i = N_C e^{(E_i - E_c)/kT}$$

$$\Rightarrow N_C = n_i e^{-(E_i - E_c)/kT}$$

$$n = n_i e^{-(E_i - E_c)/kT} \cdot e^{(E_F - E_c)/kT}$$

$$n = n_i e^{(E_F - E_i)/kT}$$

$$E_F = E_i + kT \ln \left(\frac{n}{n_i} \right)$$

$$p = N_V e^{(E_v - E_F)/kT}$$

$$p_i = N_V e^{(E_v - E_i)/kT}$$

$$\Rightarrow N_V = p_i e^{-(E_v - E_i)/kT}$$

$$p = p_i e^{-(E_v - E_i)/kT} \cdot e^{(E_v - E_F)/kT}$$

$$p = p_i e^{(E_i - E_F)/kT}$$

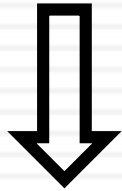
$$E_F = E_i - kT \ln \left(\frac{p}{p_i} \right)$$



Intrinsic Fermi Level, E_i

- To find E_F for an intrinsic semiconductor, we use the fact that $n = p$.

$$N_C e^{(E_i - E_c)/kT} = N_V e^{(E_v - E_i)/kT}$$

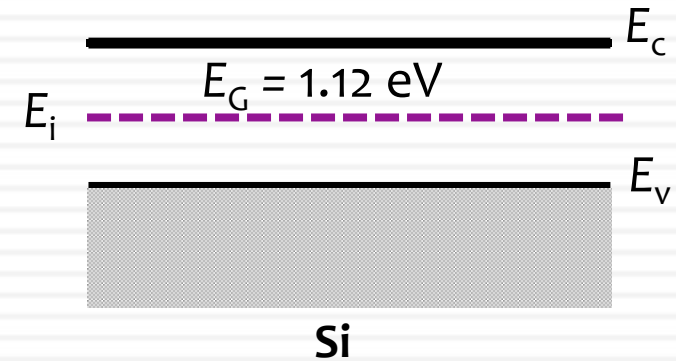


$$E_i = \frac{E_c + E_v}{2} + \frac{kT}{2} \ln \left(\frac{N_V}{N_C} \right)$$

$$E_i = \frac{E_c + E_v}{2} + \frac{3kT}{4} \ln \left(\frac{m_p^*}{m_n^*} \right)$$

$$E_i \cong \frac{E_c + E_v}{2}$$

- E_i lies (almost) in the middle between E_c and E_v





EXAMPLE: Carrier Concentrations

Question: What is the hole concentration in an N-type semiconductor with 10^{15} cm^{-3} of donors?

Solution: $n = 10^{15} \text{ cm}^{-3}$.

$$p = \frac{n_i^2}{n} \approx \frac{10^{20} \text{ cm}^{-3}}{10^{15} \text{ cm}^{-3}} = 10^5 \text{ cm}^{-3}$$

After increasing T by $60 \text{ }^\circ\text{C}$, n remains the same at 10^{15} cm^{-3} while p increases by about a factor of 2300 because $n_i^2 \propto e^{-E_g/kT}$

Question: What is n if $p = 10^{17} \text{ cm}^{-3}$ in a P-type silicon wafer?

Solution:

$$n = \frac{n_i^2}{p} \approx \frac{10^{20} \text{ cm}^{-3}}{10^{17} \text{ cm}^{-3}} = 10^3 \text{ cm}^{-3}$$



Example: Energy-Band Diagram

- For Silicon at 300 K, where is E_F if $n = 10^{17} \text{ cm}^{-3}$?

Silicon at 300 K, $n_i = 10^{10} \text{ cm}^{-3}$

$$\begin{aligned} E_F &= E_i + kT \ln \left(\frac{n}{n_i} \right) \\ &= 0.56 + 8.62 \cdot 10^{-5} \cdot 300 \cdot \ln \left(\frac{10^{17}}{10^{10}} \right) \text{ eV} \\ &= 0.56 + 0.417 \text{ eV} \\ &= \underline{\underline{0.977 \text{ eV}}} \end{aligned}$$



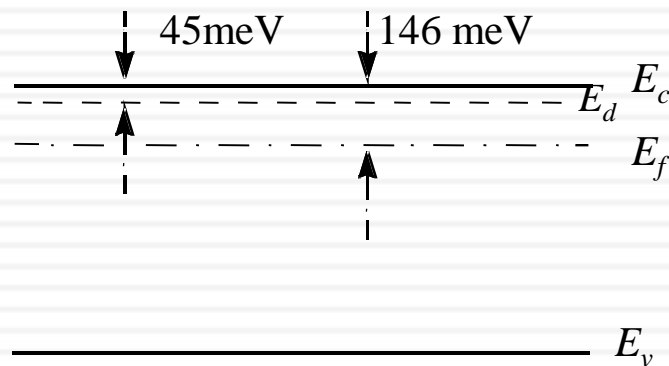
1.9 General Theory of n and p

EXAMPLE: Complete ionization of the dopant atoms

$N_d = 10^{17} \text{ cm}^{-3}$. What fraction of the donors are not ionized?

Solution: First assume that all the donors **are** ionized.

$$n = N_d = 10^{17} \text{ cm}^{-3} \Rightarrow E_f = E_c - 146 \text{ meV}$$



Probability of not being ionized $\approx \frac{1}{1 + \frac{1}{2} e^{(E_d - E_f)/kT}} = \frac{1}{1 + \frac{1}{2} e^{((146-45)\text{meV})/26\text{meV}}} = 0.04$

Therefore, it is reasonable to assume complete ionization, i.e., $n = N_d$.



1.9 General Theory of n and p (2)

Charge neutrality: $n + N_a = p + N_d$

$$np = n_i^2$$

$$p = \frac{N_a - N_d}{2} + \left[\left(\frac{N_a - N_d}{2} \right)^2 + n_i^2 \right]^{1/2}$$

$$n = \frac{N_d - N_a}{2} + \left[\left(\frac{N_d - N_a}{2} \right)^2 + n_i^2 \right]^{1/2}$$



1.9 General Theory of n and p (3)

I. $N_d - N_a \gg n_i$ (i.e., N-type)

$$n = N_d - N_a$$

$$p = n_i^2 / n$$

If $N_d \gg N_a$, $n = N_d$ and $p = n_i^2 / N_d$

II. $N_a - N_d \gg n_i$ (i.e., P-type)

$$p = N_a - N_d$$

$$n = n_i^2 / p$$

If $N_a \gg N_d$, $p = N_a$ and $n = n_i^2 / N_a$

**EXAMPLE: Dopant Compensation**

What are n and p in Si with (a) $N_d = 6 \times 10^{16} \text{ cm}^{-3}$ and $N_a = 2 \times 10^{16} \text{ cm}^{-3}$ and (b) additional $6 \times 10^{16} \text{ cm}^{-3}$ of N_a ?

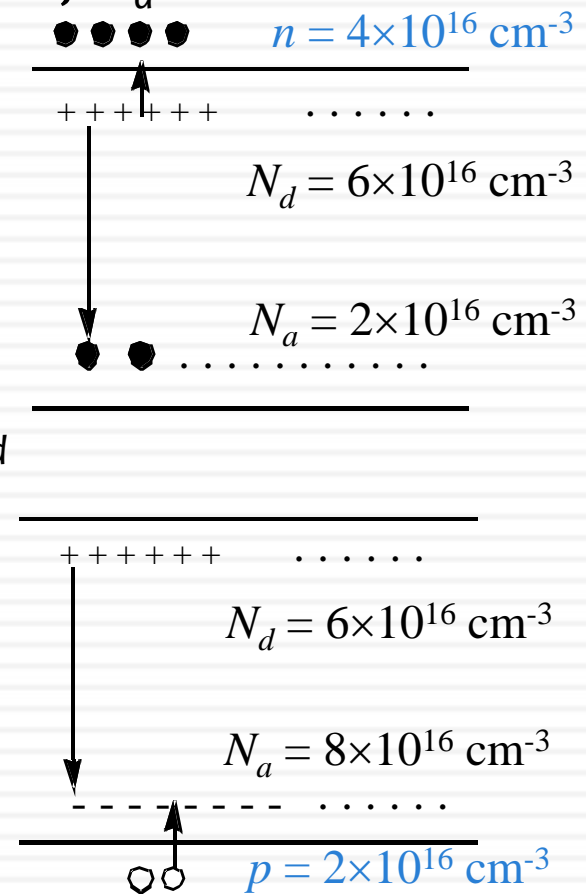
(a) $n = N_d - N_a = 4 \times 10^{16} \text{ cm}^{-3}$

$$p = n_i^2 / n = 10^{20} / 4 \times 10^{16} = 2.5 \times 10^3 \text{ cm}^{-3}$$

(b) $N_a = 2 \times 10^{16} + 6 \times 10^{16} = 8 \times 10^{16} \text{ cm}^{-3} > N_d$

$$p = N_a - N_d = 8 \times 10^{16} - 6 \times 10^{16} = 2 \times 10^{16} \text{ cm}^{-3}$$

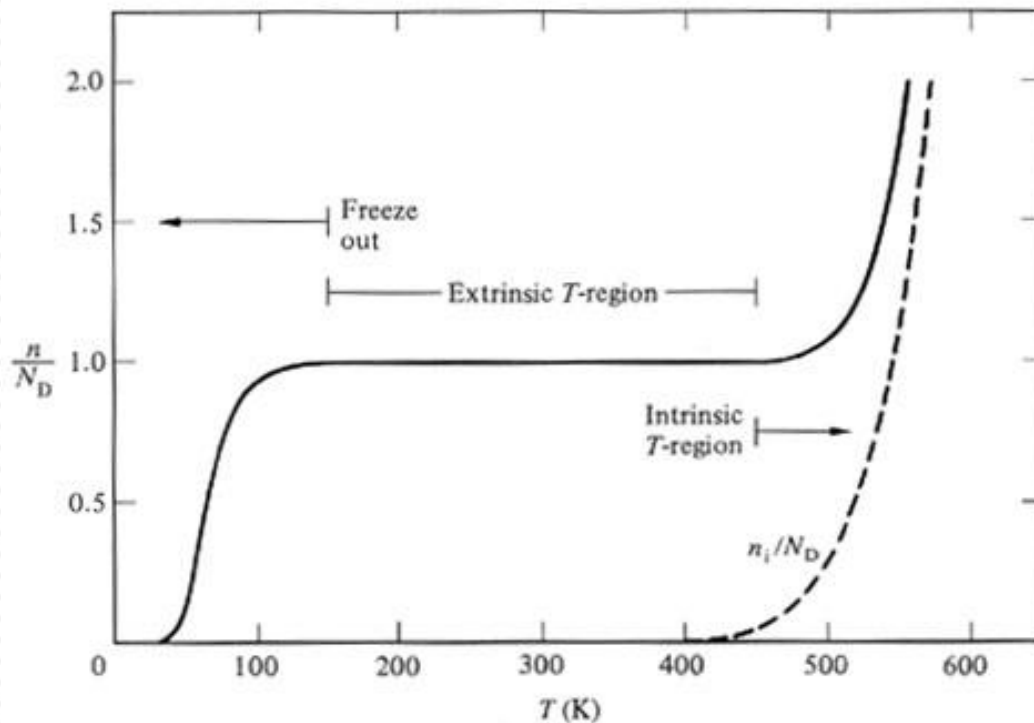
$$n = n_i^2 / p = 10^{20} / 2 \times 10^{16} = 5 \times 10^3 \text{ cm}^{-3}$$





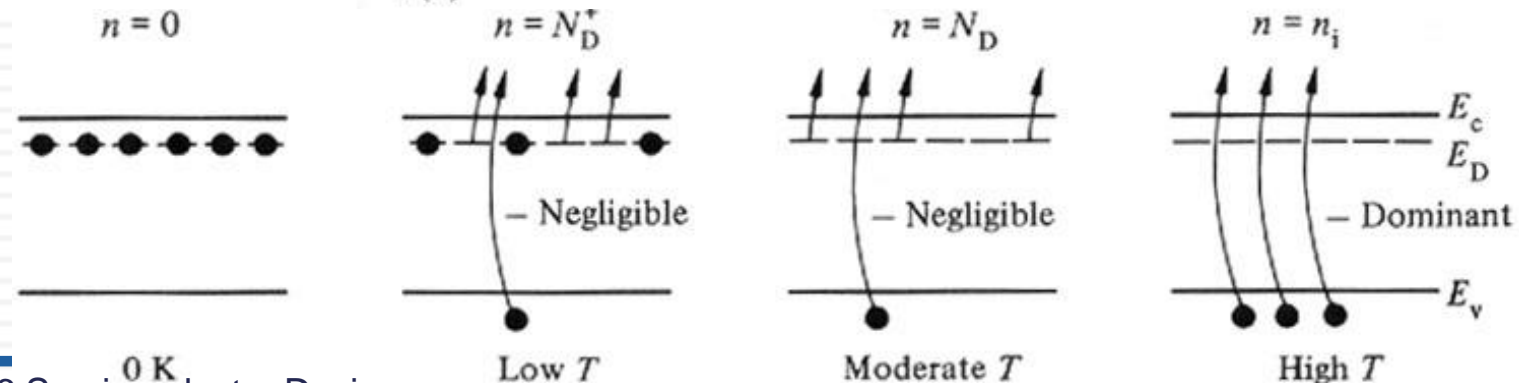
Carrier Concentration vs. Temperature

Phosphorus-doped Si



$$N_D = 10^{15} \text{ cm}^{-3}$$

- n : number of majority carrier
- N_D : number of donor electron
- n_i : number of intrinsic conductive electron





1.11 Chapter Summary

Energy band diagram. Acceptor. Donor. m_n, m_p .
Fermi function. E_f .

$$n = N_c e^{-(E_c - E_f)/kT}$$

$$p = N_v e^{-(E_f - E_v)/kT}$$

$$n = N_d - N_a$$

$$p = N_a - N_d$$

$$np = n_i^2$$