Alexandria University Faculty of Engineering **Electrical Engineering Department** Mid-term Exam, November 2013

Course Title and Code Number: Semiconductor Devices (EE336) Third Year (Communications and Electronics) Time Allowed: 45 Mins

Seat number:

Electrons

Answer All Questions in The Exam Paper:

Question 1:

Name:

(4 marks)

In the Czochralski process, a known amount of boron with a segregation coefficient $k_0 = 0.8$ is added to the melt to obtain the desired concentration. 1600

1400

1200

1000

- a) **Derive** an expression for the doping concentration in a solid (C_s/C_0) as a function of the fraction solified (M/M_0) .
- b) Assuming that a 10 kg pure silicon is used, what amount of boron must be added to get boron-doped silicon having a resistivity of 0.01 Ω -cm when one half of the ingot is grown? The density of silicon is 2.338 g/cm³ and atomic weight of boron is 10.8 g/mol.
- Mobility [cm²/Vs] 800 Holes 600 400 200 0 1.0E+14 1.0E+16 1.0E+18 1.0E+20 Doping concentration [cm⁻³]
- c) What will be the type (n or p) of a wafer produced by the above procedure? Is the dopant distribution uniform along the wafer diameter? If not, what is the region with larger impurity concentration (center or perimeter region)? And why?



جامعة الاسكندر كلبة الهندسة قسم الهندسة الكهر ىية امتحان نصف الفصل الدراسي الثاني (نوفمبر ٢٠١٣) اسم المقرر والرقم الكودي النبآئط شبه الموصلة (EE336) السنة الدر اسية الثالثة (اتصالات و الكتر ونيات)

الزمن: ٤٥ دقيقة

(15 marks)

Question 2:

(11 marks)

→ Vent

Furnance

Select the appropriate answer(s) of the following questions. You may select more than one choice if needed. Incomplete and incorrect answers will not be rewarded.

- 1. What is the fabrication process which use the following equipment shown in Figure?
 - a. Oxidation
 - b. Physical vapor deposition
 - c. Ion implantation
 - d. Chemical vapor deposition Gas inlets
 - polishing
- 2. The epitaxial layer is a:
 - a. amorphous silicon layer grown above a seed substrate wafer
 - b. single crystalline silicon layer grown above a seed substrate wafer
 - c. polysilicon layer grown above a seed substrate wafer
 - d. single crystalline silicon layer grown from a seed crystal
 - e. none of the above
- 3. The chemical vapor deposition (CVD) method for low temperature requirements is:
 - a. atmospheric-pressure CVD
 - b. plasma-enhanced CVD

- d. low-pressure CVD
- e. none of the above

Quartz tube

- c. RF sputtering
- 4. The primary semiconductor applications of physical-vapor deposition (PVD) technology are the deposition of:
 - a. polysilicon
 - b. silicon nitride
 - c. metal and compounds

- d. all of the abovee. none of the above
- 5. The implanted impurity profile of ion implantation is:
 - a. erfc distribution with a maximum at the wafer's surface
 - b. Gaussian distribution with a maximum at the wafer's surface
 - c. uniform distribution
 - d. Gaussian distribution with a maximum at a certain projection range
 - e. none of the above
- 6. In a two-step diffusion process a pre-deposition layer is first formed under a
 - a. constant-temperature condition

c. constant-pressure condition

b. constant-surface-concentration condition

- d. constant-flux condition
- e. constant-total-dopant conditio

- 7. In a two-step diffusion process, the pre-deposition layer is followed by a drive-in diffusion under a:
 - a. constant-temperature condition
 - b. constant-surface-concentration condition

- c. constant-pressure condition
- d. constant-flux condition
- e. constant-total-dopant condition

8. Assign the following lithographic projection techniques to the corresponding equipment on Figure.

- I. M:1 reduction step-and-scan
- II. 1:1 step-and-repeat

III. annual-field wafer scan



9. Sort the following fabrication steps of an n-well process according to their precedence:

- polysilicon patterning I. IV.
- II. metallization
- III. n-diffusion

- VI. contact creation

(1: ...), (2: ...), (3: ...), (4: ...), (5: ...), (6: ...) For the following equation describing the oxide thickness in thermal oxidation:

$$x = \frac{D}{k} \left[\sqrt{1 + \frac{2C_0 \kappa^2 (t + \tau)}{DC_1}} - 1 \right]$$

10. During the early stages of oxide growth, the oxide thickness varies:

- a. linearly with time
- b. exponentially with time
- c. linearly with the reaction rate
- 11. As the oxide layer becomes thicker, the oxide growth then becomes proportional to of the oxidizing time
 - a. the square
 - b. the root mean square
 - c. the square root

Good Luck

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- d. the inverse
- e. none of the above

e. linearly with

coefficient

d. linearly with the reaction rate

the

diffusion

- n-well formation
 - V. p-diffusion